October 24, 2005

Lecture 17 Microbial and Plant Toxicity

I. Overview of Microbial Biochemical Ecology

- A. Composition of Soil
 - 1. Soil essentially consists of a solid, water, and air phase (Figure 1).
 - a. The types of solids depends on the soil origin
 - 1. The solid phase consists of inorganic minerals (for example, iron oxides) and "polymers" and organic matter
 - 2. The solid components are classified by chemical composition and particle size
 - a. Clays (sheets of repeating units of Si, O or OH, Mg, Al, and several other cations)
 - 1. Particle sizes less than $2 \mu m$
 - b. Silts (particles $> 2 \mu \text{m} 50 \mu \text{m}$)
 - c. Sand (particles $>50 \mu m 2 mm$),

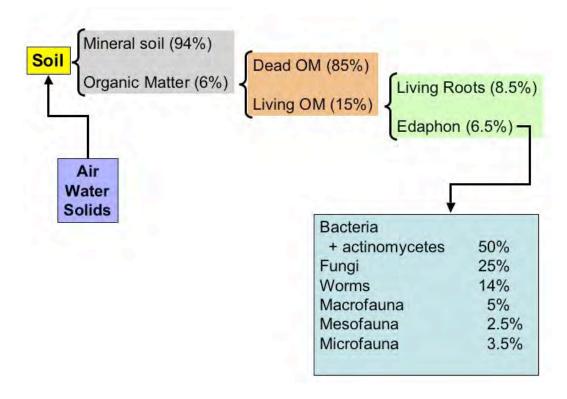


Figure 1. Composition of soil in a grassland model. Percentages represent the dry weight proportion of each component. (Adapted from figure presented in Eijsackers 1994)

b. The proportion of air and water depends on the texture (mechanical composition) of the materials and the ambient climatic conditions (i.e., has it just rained or is it droughty).

1. The solids are organized so that spaces are left around each particle (Figure 2).

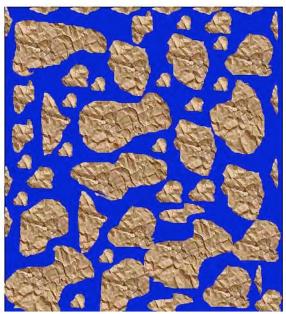


Figure 2. Packing of soil particles determines the amount of pore space and thus the amount of water the soil can hold when all the spaces are filled. Spaces partially filled with water are filled with air.

- a. These spaces form a network of continuous and discontinuous pores that are filled with either water or air.
- 2. The maximum amount of water that the soil can hold depends on the texture of the soil (percentage composition of clay, silt, and sand) as well as the type of solids and amount of organic matter.
 - a. When all the pores are filled with water, the soil is described as "saturated". Further additions of water will displace water to areas of unsaturation (pores contain air and some water).
- 2. Organic matter (OM) content is normally small by proportion to the inorganic components (for example, the clay fraction);
 - a. A soil with 6% organic matter (OM), for example, would be considered to have a moderately high content of OM.
 - b. Soils of east central WA tend to have <2% OM and very little clay.
 - c. Organic matter can be further divided into dead and living OM (Figure 1).

II. Concerns over microbial toxicity stem from the central role of soil microorganisms in biogeochemical cycling and soil fertility

- A. Biogeochemical cycling mainly involves the cycling of carbon, nitrogen, phosphorus, and sulfur from inorganic to organic forms, and back again to inorganic forms. (See Figures 3, 4, and 5)
- B. All plant life depends on bioavailability of nutrients, especially nitrogen.

- 1. Microorganisms decompose organic matter and recycle C, N, P, and S by turning it from organic species into bioavailable inorganic forms that plants can absorb and metabolize.
 - a. Thus adverse effects on microbial functional diversity can reduce soil fertility and thus plant biomass.

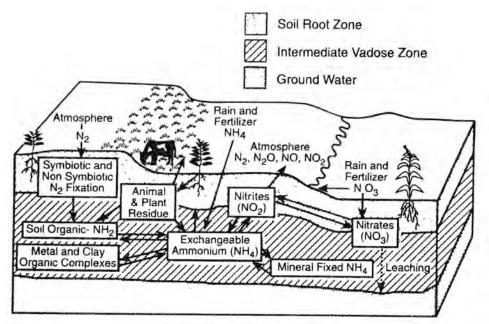


Figure 3. Cycling of nitrogen in the environment (figure from Rechcigl, 1995, Soil Amendments & Environmental Quality, CRC Press). Note that nitrogen goes through a series of oxidations and reductions. Nitrogen species are found in air, water, and soil.

NH₃

pH dependent equilibrium

N (organic)

NH₄
$$\stackrel{4}{=}$$
 NO₂ $\stackrel{4}{=}$ NO₃

N₂ $\stackrel{4}{=}$ NO₂

N₂ $\stackrel{5}{=}$ N₂O

Figure 4. Nitrogen reactivity in the biogeochemical cycle. Numbers refer to specific oxidation or reduction reactions. (1) <u>Assimilation of inorganic N</u> (mainly NH₄⁺, which is in a pH dependent equilibrium with NH₃, and NO₃⁻) by microorganisms and plants to form organic N such as amino acids (which become proteins) and nucleotide bases (which become "polymerized" into DNA, RNA). Also known as immobilization. The metabolic pathway is called the glutamine synthase-

synthetase pathway; **(2)** <u>Heterotrophic conversions</u> involving the transfer of organic N among organisms; **(3)** <u>Mineralization (or ammonification)</u> involves degradation of organic matter containing N to NH₄⁺ by bacteria and fungi; **(4)** <u>Nitrification</u> involves conversion of NH₄⁺ (ammonium ion) to NO₂⁻ (nitrite) and NO₃⁻ (nitrate); the reaction is microbial-mediated oxidation (by nitrifying bacteria); **(5)** <u>Denitrification</u> involves reduction of nitrate to nitrite; nitrite can be further reduced to gaseous N₂O and N₂ **(6)** <u>Biological Nitrogen Fixation</u> involves direct absorption of atmospheric N₂ to form NH₄⁺

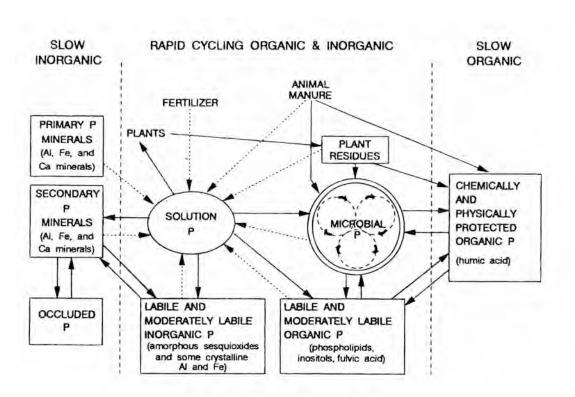


Figure 5. Phosphorus cycling in the environment. Note that there is no air phase in the cycle as there is for nitrogen cycling.

- C. Distribution of Microorganisms in the Soil (Table 1)
 - 1. Most microorganisms, and thus microbial activity most important in biogeochemical cycling, are near the surface of the soil.

Table 1. Distribution of microbial flora in soil (numbers x 10³/gram soil) (Eijsackers 1994)

Depth (cm)	Aerobic Bacteria	Anaerobic Bacteria	Actinomycetes	Fungi	Algae
3-8	7800	1950	2080	119	25
20-25	1800	379	245	50	5
35-40	472	98	49	14	0.5
65-75	10	1	5	6	0.1
135–145	1	0.4		3	

- D. Attention to microbial toxicity has been focused on essentially two activities:
 - 1. Disposal of wastes;
 - a. Waste disposal activities, whether legally in a landfill, or inadvertently through spills of material, create very high concentrations.
 - 1. Given "dose makes the poison", the probability of causing microbial toxicity and thereby reducing microbial activity faces a high probability of occurrence.
 - 2. Pesticide application;
 - a. The application process will always produce residues on soil regardless of the intended target.
 - 1. Often, the intended target may be soil directly:
 - a. Weed control had conventionally relied on pre-emergent use of herbicides (i.e., herbicides applied directly to the soil before the crop seedling emerges)
 - 1. Modern weed control mostly uses post-emergent herbicides that are applied to growing weeds in the crop (i.e., the herbicide is applied over the top of the emerged crop or to the side of the emerged crop). Obviously, post emergent herbicides are not toxic to the crop at the recommended rates of application.
 - 2. Large acreages of farmland, mainly in the Corn Belt are treated with either atrazine (corn: US total acreage was ~81 million acres in 2004) or glyphosate (specifically on genetically engineered herbicide resistant soybeans: US total acreage was ~75 million acres in 2004) (USDA 2005)
 - 2. Control of soil insects relies on the use of soil insecticides
 - a. Soil insecticides used in corn production, the biggest consumer of insecticides from the viewpoint of total poundage, are actually applied in 18-cm bands over the seed row rather than broadcast throughout the field.
 - 3. Fumigation with methyl bromide, 1,3-D (dichloropropene), and metamsodium (which quickly hydrolyzes to the toxicant, methyl isothiocyanate)
 - a. Fumigants are broad-spectrum pesticides that kill seedling plants, nematodes, pathogenic fungi, as well as depress microbial biochemical activity.
 - 4. A number of fungicides are used to treat foliar pathogens, many of which are fungi.
 - a. Thus, inadvertent deposition of fungicides on soil could adversely affect soil fungi

III. Measuring Microbial Toxicity

- A. "In Vitro" Tests
 - 1. Microorganisms can be cultured in liquid broth or on agar.
 - a. Population size can be established as turbidity units (if in liquid culture) or as colony forming units (if on agar plates).

- b. Can add toxin to the culture and determine concentration related to effect on microbial numbers.
- 2. Can examine rates of respiration or other specific enzyme reactions.

B. "In Vivo" Tests

- 1. Whole soils can be tested for microbial activity, including
 - a. Respiration (overall measurement for microbial activity); measured as the activity of dehydrogenase enzyme;
 - b. Evolution of carbon dioxide (as a surrogate for microbial biomass and potential for mineralization of organic carbon);
 - c. Enzyme specific reactions for processing nitrogen;
 - 1. Nitrification
 - 2. Nitrogenase (nitrogen fixation)
 - 3. Urease
 - d. Phosphatase activity (decomposition of organic phosphorus to phosphate).
- 2. Microbial groups having specific biochemical functions can be enumerated using Most Probable Number techniques.
 - a. For example, one general isolation would be aerobic and anaerobic microorganisms.
 - b. Microbes capable of reducing nitrate to nitrite could be isolated from those that oxidize nitrite to nitrate.

C. "Ecological" Tests

- 1. Decomposition of litter
 - a. Leaf disks or material placed in mesh bags and buried in soil;
 - b. Monitor disappearance of biomass over time (by visual inspection and by weighing);
 - c. Some discrepancy between field studies and lab studies have been noted; the lab studies do not seem predictive of the field studies
 - 1. de Jong (1998); Abstract:
 - a. "A method has been developed for assessing the side effects of fungicides on decomposition using litterbags. Twenty dried leaf disks of Chinese cabbage, Brassica oleracea, were placed in litterbags (20x20 cm) made of nylon fabric (25 μ m mesh). The litterbags were laid in a treated plot and 5, 10, 20, and >200 m from the treated plot and covered with 1 cm of standard soil. After 1 week, the dry weight of the leaf disks was determined. In an iterative procedure the method was optimized and significant differences of 5% in decomposition rate between litterbags were found using the final method presented here. In and around agriculturally managed plots, negative effects on decomposition of two fungicides could be traced. Effects on decomposition were found after application of captan (in fruit, 1.5 kg a.i./ha) and maneb (in potatoes, 3 kg a.i./ha) up to 10 m from the treated plot, at moderate wind speeds (35 m/s). In one case the concentration of captan in the soil covering the litterbags was measured. A negative correlation was established between captan concentration and decomposition. These experiments were repeated in an experimental plot under more controlled conditions with respect to the amount and time of application. The results of the field experiments could not be reproduced on the experimental plot. Additional experiments with soil fungi in agar again demonstrated a significant effect of exposure to captan, however. Given the results, the method using litterbags requires further development before a standard field trial can be designed. Given the knowledge gained with respect to the key factors involved a bioassay using soil fungi seems to have high potential, however."

IV. Case Studies of Microbial Toxicity

- A. Effects of Pesticide Applications on Microbial Activity
 - 1. Studies of Tu (1981a; 1981b) show that at normal application rates and doubled rates, any changes in various measures of microbial activity (when they occur) are temporary and the soil recovers quickly.
- B. There has been concern about the long term use of pesticides and its effect on soil microbial activity and therefore fertility.
 - 1. Research at the Rothamsted Research Station in the U.K. has been conducted since the late 1800's and represents the longest running controlled experiments of agriculture worldwide.
 - 2. A study started in the early 1970's has been assessed for the effect of pesticides on soil fertility and crop health. No long term effects were noted as concluded in the abstract below (Hart and Brookes 1996). Nor was there any effect on crop productivity (Bromilow et al. 1996).
 - a. Hart and Brookes (1996) Abstract:
 - "The effects of 19 years of cumulative annual field application of five pesticides (benomyl, chlorfenvinphos, aldicarb, triadimefon and glyphosate), applied at, or slightly above, the recommended rates in 25 combinations, on soil microbial biomass and the mineralization of soil organic matter were investigated. Soil samples were taken 1 month after the application of benomyl, chlorfenvinphos and aldicarb in April 1992, and again in October 1992, 1 month after the application of triadimefon and glyphosate. The addition of aldicarb caused a significant increase of 7-16% in soil microbial biomass carbon (biomass C), an effect which appeared to be persistent. This effect of aldicarb was not reflected in the mineralization rate of soil organic C, possibly because the measurements of CO2 evolution showed a greater variation than those of biomass C. Measurement of microbial biomass activity by the substrate-induced respiration method also gave much less precise results than measurements of biomass C by fumigation-extraction. The mineralization of soil organic N to ammonium and then nitrate was mostly unaffected by the pesticide treatments. In the autumn-sampled soil, there was significantly less NH4-N in the aldicarb-treated soil. It is possible that this was due to immobilization by the increased microbial biomass in these treatments, and did not represent a loss to the soil system. The continuous use of these pesticides, either singly or in combination, therefore had no measurable long-term harmful effects on the soil microbial biomass or its activity, as assessed by C or N mineralization."
 - 2. Note that in the experiment by Hart and Brookes, microbial biomass and bioactivity were measured in the 19th year in the separate experimental plots. Each plot was kept separate and treated either singly or with a combination of herbicides/insecticides. In the spring and autumn of the last year of treatment, measurements were made. Figure 6 shows a typical result for microbial biomass measurements in the plots that received each of the denoted herbicides.

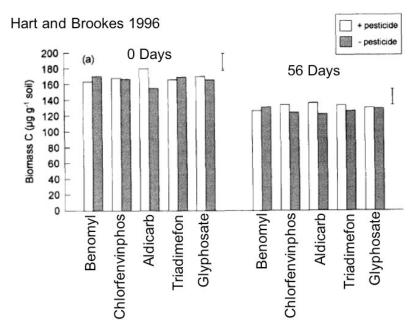


Figure 6. Effect of herbicide application on microbial biomass in long-term pesticidetreated plots at Rothamsted Experiment Station in the U.K. (Hart & Brookes 1996).

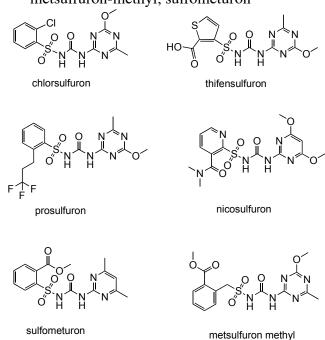
C. Effects of Waste Spills

- 1. Dzantor and Felsot (1991)
 - a. Simulated spills of the herbicide alachlor significantly reduced microbial biomass and bioactivity as measured by the Most Probable Number method and dehydrogenase activity.
 - b. At normal rates of application, no effects on microbial biomass or activity were observed.

V. Plant Toxicity

- A. Similar to concerns about microbial toxicity, agricultural activities leading to exposure by pesticides (especially herbicides) at the field margins or alternatively due to longer distance movement in drift have been of concern for effects on productivity of nontarget plants.
 - 1. Another concern has been the re-establishment of plant populations on former waste sites.
- B. With regard to agricultural chemicals, herbicides are of most concern.
 - 1. Intended to kill plants, herbicides can drift during spraying and potentially affect nontarget plants around field margins.
 - 2. Under certain conditions, herbicide residues can move much farther distances than the field margin and cause morphological effects that could be economically injurious
 - a. A good example is the extraordinary sensitivity of grapes to 2,4-D.
 - b. 2,4-D has been historically used to kill broadleaf weeds in cereal crops.
 - c. Off-target movement of 2,4-D, either as direct drift and movement in wind, or as volatilized residues and movement in inversions, can cause injury to grapes that are in the path of the moving air masses.

- 1. Many times, the injury is not damaging to yield, but rather causes notable morphological symptomology.
- 3. In addition to off-target movement during spraying, some herbicides can persist through the growing season through the beginning of the next season. This phenomenon and associated effects are called "carry-over".
 - a. When crops are rotated, these new crops may be more very sensitive to the herbicide at low levels, and thus suffer injury leading to productivity loss.
- C. **Selected Examples of Herbicides and Mode of Action** (those that have been most studied from an environmental toxicology perspective)
 - 1. The vast majority of herbicides have mechanisms of toxic action that is specific to plants.
 - a. However, note that for some herbicides, the mechanism could potentially affect bacteria.
 - 2. **Mode of action definitively known**; likely related to one enzyme or receptor interaction
 - a. Sulfonylurea (SU) herbicides and imidazolinone herbicides inhibit acetolactate synthase (ALS), an enzyme only occurring in bacteria and plants
 - 1. ALS is a key enzyme on the metabolic pathway for synthesis of branched chain amino acids (valine, leucine, isoleucine).
 - 2. These herbicides are effective at use rates of grams per hectare (ounces per acre).
 - a. These herbicides have biological activity when taken up from soil and by direct contact with foliage.
 - 3. Examples:
 - a. SU herbicides
 - 1. chlorsulfuron, thifensulfuron, prosulfuron, nicosulfuron, , metsulfuron-methyl, sulfometuron



- b. Imidazolinone herbicides
 - 1. imazapyr
 - 2. imazethapyr
 - 3. imazaquin

- b. Glyphosate
 - 1. Inhibits the enzyme 5-enolpyruvylshikimic acid-3-phosphate synthase (EPSPS) in the shikimate acid biosynthetic pathway for synthesis of aromatic amino acids (tryptophan, phenylalanine, tyrosine)
 - a. Only bacteria and plants have this enzyme (i.e., the shikimate acid pathway)
 - 2. Glyphosate is only active when it contacts the leaf surface.

glyphosate

- a. It is a true systemic, meaning that it can move in the phloem down to the roots.
- b. No activity when residues contact soil; due to lack of bioavailability to plant from soil.
- 3. Rates of glyphosate application are in kg/ha (pounds per acre).
- 4. Soybeans, cotton, and canola (called rape in Europe) have been transformed via genetic engineering with an EPSP gene resistant that is resistant to binding by glyphosate, making the crop resistant (for further information see Felsot 2000).
- c. Auxin mimics (includes phenoxyacetic acids, such as 2,4-D and mecoprop; the benzoic acid dicamba; and the picolinic acids picloram, triclopyr, and clopyralid)
 - 1. Mimic the natural plant growth hormone indole-3-acetic acid (auxin);

indole-3-acetic acid (auxin)

2. Rates of application range from fractions of a kg/ha to several kg/ha.

- d. Photosynthesis inhibitors
 - 1. Includes a large number of herbicides, but the most studied one is atrazine;

- 2. Atrazine, which is a symmetrical triazine compound, has become the subject of an enormous number of papers because it is widespread in the environment, especially in aquatic habitats.
- 3. Atrazine was one of the first broadleaf herbicides that was useful in corn production.
 - a. However, as a result of its widespread use and rates of application of several kg/ha, it is the most frequently found pesticide residue in water worldwide.
- 4. Simazine used to be used frequently in viticulture, but most growers now just treat the rows with a band of glyphosate.
- 5. In the potato growing regions of the PNW, metribuzin (an asymmetrical triazine herbicide) was frequently applied to bare soil at kg/ha rates.
- 3. Non-specific mechanisms
 - a. Acylanilide (a.k.a. chloroacetamides): includes alachlor and metolachlor

1. These herbicides, which have been used quite heavily in corn production, are also frequent contaminants of water (but not to the same degree as atrazine)

- 2. The specific of causing phytotoxicity remains unknown, but this group seems to affect lipid metabolism.
- 3. Applied directly to soil at rates of kg/ha.
- b. Bipyridinilium Herbicides (Paraquat and Diquat)

- 1. These herbicides, which are cationic (positively charged), cause a reaction that enhances free radical generation in the cell. Free radicals react with all other biochemicals and cause destruction of the cell via oxidative mechanisms.
- 2. Bipyridinilium herbicides can only effect toxicity when in direct contact with the plant surface.
- 3. Rates of application tend to be kg/ha.
- c. Dinitroanilines (trifluralin; oryzalin; pendimethalin)

- 1. Adversely affect root elongation by inhibiting mitotic spindle formation; thus, cells cannot divide.
- 2. Applied at rates of sub kg/ha.

VI. Assessing Plant Toxicity

- A. Presently, EPA requires that tests for effects of pesticides on plants be conducted on a few surrogate species including alfalfa and soybeans. Basically there are three types of tests that all pesticides are subject to regardless of their type (i.e., herbicide insecticide, etc.).
 - 1. Germination tests
 - a. Seeds are treated with the pesticides and then examined for germination potential.
 - 2. Root elongation tests
 - a. Seeds are treated with the pesticide or the newly sprouted hypocotyl is treated.
 - b. Roots are examined for growth (elongation) potential.
 - 3. Growth response
 - a. Plants are potted and then sprayed with a pesticide.
 - b. Growth is measured both as biomass (just clip the plant above the soil surface, dry it at a specified temperature, and weigh it) or as height above the soil surface.
 - c. Plot concentration relative to biomass and estimate concentration causing 25% reduction in growth (called the GR25) (Figure 7).

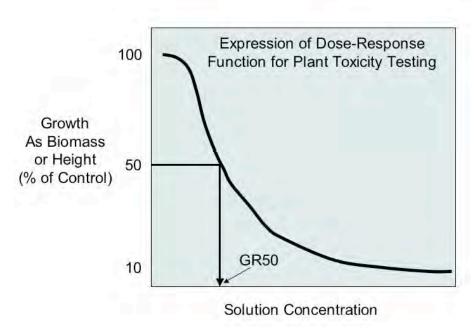


Figure 7. Expression of dose-response function for plant toxicity testing.

B. Ecological Observations

- 1. Field observations of phytotoxic effects of pesticides are possible but the problem is the effects are likely to be localized.
- 2. However, productivity as measured by chlorophyll A concentration in algal cells, or alternatively, gross photosynthetic activity can be measured under laboratory conditions.

- a. Photosynthesis can be measured in the field using portable instrumentation.
- 3. Effects on productivity of aquatic plants can be viewed as a surrogate for higher level ecological effects.
 - a. Recruitment in to aquatic habitats is likely to be more restricted than recruitment of plant species on the land.
 - b. An example of using productivity of algae (or phytoplankton) as a regulatory endpoint is EPA's determination of the most sensitive endpoint for the herbicide atrazine in its ecological risk assessment as part of the pesticide re-registration process. (The web site URL for registration eligibility decision documents, known as REDs, can be found at http://www.epa.gov/pesticides/reregistration/status.htm; just click on the atrazine link and you will be taken to a number of documents regarding EPA assessment of atrazine risk. You'll be looking for the document known as EFED (Ecological Fate and Effects Division).
 - 1. The most sensitive toxicological endpoint is an adverse effect on phytoplankton productivity at a NOAEL of $2.3 \mu g/L$.
- C. EPA is considering whether to revise its plant toxicity testing requirements to include effects on plant reproduction.
 - 1. The concern over plant reproduction came out of research with the sulfonylurea (SU) herbicides, essentially conducted by one group of researchers at the EPA Corvallis Lab and Oklahoma State University (Fletcher et al. 1993, 1995, 1996).
 - a. This research was motivated by complaints of herbicide drift from wheat fields on the Horse Heaven Hills above the Badger Canyon in Benton Co., WA.
 - 1. The herbicides involved were sulfonylureas, and at the time of the complaints they were applied by aerial spraying
 - 2. A couple (literally) of residents in the Badger Canyon complained to both the Washington Dept. of Agriculture and the EPA that the productivity of their crops (generally alfalfa and cherries) were being adversely effected.
 - 3. One of the tell-tale signs of low levels of SU exposure is an isolated diffuse yellow spot on the foliage (Felsot et al. 1996a).
 - a. Cherry leaves, which unfold in April (i.e., comparatively early in the growing season) and have a very shiny waxy surface, are particularly good bioindicators of exposure to low levels of SU herbicide residues.
 - b. Working with cherry trees, pea plants, and soybeans, Fletcher et al. reported they saw injurious effects on plant reproduction (for example, increased incidence of flower abortion) from low levels of SU herbicides without seeing any concomitant injury on leaves.
 - 1. Fletcher et al. claimed that the effect on reproduction would only occur if the plant was in a specific growth stage.
 - 2. In contrast to the Fletcher et al. observations, Bhatti et al. (1995) observed no effect of chlorsulfuron on cherry fruit production in the absence of easily observable leaf injury.

- a. Furthermore, comparatively high concentrations of chlorsulfuron were required to affect reproduction (as measured by fruit abortion).
- 3. To determine the likelihood that spray applications in the Horse Heaven Hills might be directly affecting plants in the Badger Canyon, sentinel plants were set up at different locations and exchanged on a weekly basis.
 - a. Plants were returned to the WSU-Prosser research station (Irrigated Research and Extension Center, a.k.a. WSU IAREC) and examined for tell-tale signs of morphological changes indicative of exposure to SU herbicides, 2,4-D (or similar auxin agonist herbicides), paraquat, and glyphosate.
 - b. To make a long story short (see Felsot et al. 1996b for the full picture), the appearance of leaf symptomology characteristic of exposure to either SU or 2,4-D residues was not correlated with times of spraying.
 - 1. A general atmospheric circulation and deposition of residues, perhaps associated with early morning very light drizzles, was more likely the route of exposure and not direct spray drift.
 - 2. Furthermore, given the nature of the morphological phenomenon on leaf surfaces, productivity of plants was unlikely to be affected.
- D. Plants of "conservation" interest (i.e., native plant species) adjacent to farm fields have been studied in the U.K. (Marrs et al. 1991a,b).
 - 1. Marrs et al. simulated drift of herbicides adjacent to an agricultural field. He would set out potted plants of interest and determine the effects on growth, appearance, and fecundity.
 - 2. The age of plants definitively affected susceptibility.
 - 3. Depending on the species saw varied affects, including fecundity
 - a. Made observations at a distance of 8 m from the spray source.

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